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2 **After-coal diamonds: an enigmatic type of impact diamonds**
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17 **Abstract**

18 Impact diamonds were discovered in the 70s and usually accepted as being paramorphs
19 after graphite, resulting in grains of extremely high mechanical quality. A diffusion-less
20 mechanism for the graphite-to-diamond transition under huge pressure has been experimentally
21 realized and theoretically explained. Besides, another type of impact product has received much
22 less attention, namely diamonds formed after coal as a result of the impact. Here we describe
23 after-coal impact diamonds from the giant Kara astrobleme (Pay-Khoy, Russia), which resulted
24 from a large asteroid impact about 70 Ma ago. The impact created a large number of unusual
25 impact diamonds, which are described here for the first time using high-resolution techniques
26 including visible and UV Raman spectroscopy, scanning electron microscopy (SEM), atomic
27 force microscopy (AFM) and transmission electron microscopy (TEM). Two main varieties of
28 after-coal diamonds occur: micrograined (sugar-like, subdivided into coherent and friable) and,
29 as a new type, paramorphs after organic relics. After-coal diamonds differ from after-graphite
30 impact diamonds by the texture, the absence of lonsdaleite, a micro- and nanoporous structure.

31 The sugar-like variety is presented by tightly aggregated well-shaped single nanocrystals. The
32 after-organic diamond paramorphs are characterized by a perfect-preserved relict organic
33 morphology, sub-nanocrystalline–amorphous sp^3 -carbon (ta-C) nanocomposites and specific
34 properties. Based on the description of after-coal diamonds, we propose a new, polystage
35 formation mechanism: high-velocity coal pyrolysis with hetero-elements removal followed by
36 pure carbon crystallization with local diffusion process resulting diffusion-limited crystallization.
37 Also we propose similarity of the aftercoal diamonds features to carbonado.

38

39 **Key-words:** astrobleme, impact diamond, coal, diamond paramorphs, after-coal
40 transformation, carbon, carbonado, shock metamorphism.

41

42 **1. Introduction**

43 Impact diamonds have been attracting significant attention since their initial discovery.
44 There are numerous publications devoted both to their formation mechanisms and to the
45 formation of the related impact craters (Bundy & Kasper, 1967; Masaitis et al., 1972, 1990,
46 1999; Sokhor et al., 1973; Shafranovsky, 1985; Val`ter et al., 1992; Decarli, 1995; Langenhorst
47 et al., 1998; El Goresy et al., 2001, 2003; Langenhorst, 2002; Vishnevsky, 2007; Kvasnytsya &
48 Wirth, 2013; Smith & Godard, 2009; Ohfuji et al., 2015). Impact diamonds display exceptional
49 mechanical properties such as extreme hardness and can occur in large deposits with high
50 concentrations, such as in the Popigai astrobleme (Masaitis et al., 1972, 1990, 1999). It is usually
51 accepted that impact-diamond formation is the result of a solid-state transformation (diffusion-
52 less process) of graphite to diamond at high pressures (≥ 30 GPa). This transformation process
53 has been supported by experimental and theoretical studies (*e.g.* Lonsdale, 1971; Kurdumov et
54 al., 2012; Xie et al., 2014). Diamond production under shock pressure is known to be also
55 possible from differently ordered carbons and coal, through a diffusion-limited reconstructive
56 mechanism (Borimchuck et al., 1991). However, the detailed mechanism of diamond formation

57 from coal is completely unclear. Besides, experiments using Kara sedimentary organic material
58 as precursor (Korochantsev, 2004) suggest that the shock pressure for the bituminoid-to-diamond
59 transformation has to be essentially higher than that for the graphite-to-diamond transition.

60 Indeed, in the experimental study of such a transformation, diamonds were not produced from
61 bituminoid even under 60 GPa, only nanocrystalline graphite was formed (Martirosyan, 2014).

62 Although impact after-coal diamond products are known from nature, information about
63 the natural transformation of non-graphitic carbon to diamond is scarce. A few detailed studies
64 devoted to such type of impact diamond and their occurrence are related to the Kara astrobleme
65 (Yezerkiy, 1982, 1986; Mashchak, 1990, 1991; Shishkin et al., 2012). The Kara astrobleme
66 (about 65 km in diameter) resulted from the impact of a large asteroid about 70 Ma ago
67 (Masaitis, 1975; Nazarov et al., 1989, 1990, 1991, 1992; Koeberl et al., 1990; Kolesnikov et al.,
68 1990; Mashchak, 1990, 1991; Tieloff, 1998; Shishkin et al., 2012) and is the only known
69 astrobleme in the northeastern European part of Russia that has been described as containing
70 diamonds formed essentially from coal rather than from graphite (Yezerkiy, 1986). According
71 to recent data (Shishkin et al., 2012), the prospected resources of impact diamond in the Kara
72 astrobleme are 200 kg/km².

73 Here we characterize for the first time a variety of after-coal carbon phases from the Kara
74 structure by detailed nanostructural analysis using high-resolution methods, including visible and
75 ultra-violet Raman spectroscopy, scanning electron microscopy (SEM), atomic force microscopy
76 (AFM) and transmission electron microscopy (TEM). The observations reveal unusual
77 nanostructural features and suggest a specific mechanism of Kara diamond formation.

78

79 **2. Kara astrobleme target composition**

80 **2.1. Geological setting**

81 The twin impact craters Kara and Ust`-Kara are located in the northeastern European part
82 of Russia belonging to the northeastern section of the Pay-Khoy Ridge structure (Fig. 1). The

83 craters have diameters of 65 and 25 km, respectively. The Kara astrobleme is fully on land while
84 Ust`-Kara partly extends just beyond the coastline of the Kara Sea, and is generally under water
85 at present. The Kara astrobleme is obvious today as a depression in the landscape, and is well
86 expressed in gravitational- and magnetic-field surveys (Mashchak, 1990, 1991; Shishkin et al.,
87 2012).

88 Past studies on the origin of the Kara astrobleme have been controversial. Until the
89 1970s, the origin of the Kara depression was assigned to a volcanic eruption. In 1970, P.S.
90 Voronov wrote in a media press about the impact hypothesis, which was immediately supported
91 by V.L. Masaitis in 1971 and further developed by industrial geologists headed by M.A. Maslov
92 and G.Ya. Ponomarev (Mashchak, 1990, 1991). Subsequent scientific studies have been carried
93 out since the 1980s by, *e.g.*, Yushkin & Lysyuk (2001) and Mal`kov & Andreichev (2010). Most
94 recently, the full range of silica polymorphs, including high-pressure coesite and low-pressure
95 post-impact phases, was described in Kara impactites by Lutoev & Lysyuk (2015).

96 The Ust`-Kara astrobleme has been known since 1976, when its nature as an independent
97 impact crater, of contemporary origin with the Kara astrobleme, was determined (Mashchak
98 1990, 1991). At present it is accepted that the Kara and Ust`-Kara astroblemes resulted from the
99 same impact event produced by a single bolide that broke up before contact with the Earth's
100 surface.

101

102 **2.2. Geological composition of the sedimentary target**

103 The geological composition is described here after Mashchak (1990) and Shishkin et al.
104 (2012), who studied the region in detail and ascribed the Kara region to the Pay-Khoy subzone of
105 the Zilairo-Lemvinskaya zone on the basis of the bathyal sediments present in the
106 Sylovayachinskaya (D₃-C₁), Karskaya (C₁) and Karasilovaya (C₂-P₁) units.

107 According to these authors, the target in the Kara and Ust`-Kara astroblemes region
108 consists of two structural levels – Late Proterozoic (bottom) and Paleozoic (upper) sediments

109 (Fig. 1). The upper structural level consists of Ordovician-Carboniferous and early Permian
110 sediments. The lower one consists of mica-clay, siliceous and actinolite-bearing phyllites with
111 lenses of metamorphosed rhyolites and tuffs, with a total thickness of more than 6 km. At the
112 center of the Kara astrobleme these lithologies occur at a depth of about 500 m. The rocks are
113 folded as in the Pay-Khoy anticlinorium system.

114 The upper structural level is comprised of two subunits. The bottom sublevel consists of
115 about 3.5 km thick Ordovician-Carboniferous sediments, where the Ordovician sediments
116 consist of clay-siliceous, mica-siliceous and carbonate-clay shales, clay and mica-containing
117 limestones measuring 1.2-1.5 km in thickness. The S–D₁ sediments are undivided, have a
118 thickness of about 370 m, and consist of black coal-clay, coal-siliceous, coal-carbonate-siliceous
119 and graphitic-siliceous shales with sublayers of quartz-containing limestones. The D₂–D₃
120 sediments of 700–900 m total thickness are pure or carbonate-containing sandstones, coal-
121 siliceous, siliceous and clay shales, jasperoids, dolomitised and marmorised limestones. The
122 Carboniferous system has a total thickness about 760 m. It is composed mainly of coal-clay,
123 coal-siliceous and clay-siliceous shales, of coal-, clay- and silica-containing limestone for the
124 bottom level (C₁ sediments). The Middle and Upper Carboniferous rocks are made up of 170 m
125 thick coal-siliceous and chlorite–sericite siliceous shales. Both Devonian and Carboniferous
126 sediments are intensely fragmented.

127 The upper sublevel is represented by structurally discordant early Permian sediments with
128 a maximum thickness of 2.1 km, widely distributed in the north-eastern part of the Pay-Khoy
129 Ridge. It is characterized by polymict sandstones, siltstones, sandstones, clay shales with
130 sublayers of limestones, coals and conglomerates, reaching a thickness of up to 2 km. This
131 structural level is less deformed than the bottom sublevel, and shows simple folding.

132 The Paleozoic sediments are overlain by Pliocene-Quaternary sediments with a typical
133 thickness of 5–10 m, rarely up to 60 m, which widely cover the area including the Kara
134 astrobleme region.

135 It is clear from this lithological description that the Kara target contains a significant
136 quantity of coal-containing sediments, amounting to hundreds of meters in thickness. This is the
137 most important feature of the giant astrobleme, in which the carbonaceous matter was the source
138 of carbon that resulted in diamond formation during the giant Kara impact event. This event
139 occurred around the Cretaceous–Paleogene boundary, with the estimated age ranging from 55
140 to 72 Ma depending on the methods used (*e.g.* Mashchak, 1990, 1991; Mal'kov & Andreichev,
141 2010). According to the most recent ^{40}Ar - ^{39}Ar determination on impact-melt rocks, the age
142 corresponds to 70.3 ± 2.2 Ma (2σ) (Trieloff et al., 1998).

143

144 **2.3. Impact rocks of the Kara astrobleme**

145 According to Shishkin et al. (2012), a specific Kara complex of rocks connected to the
146 impact event (impactites) can be divided into three units, including (from bottom to top)
147 allogenic lithic breccias, suevites, and tagamites, depending on the proportion of melt (impact
148 glass/crystallized impact melt) in the brecciated rocks (<15, 15–65 and >65 vol%, respectively).
149 Here we provide a brief description of the impact units after Machshak (1990) and Shishkin et al.
150 (2012).

151 The *allogenic lithic breccias* are characterized by dislocated impact-metamorphosed rocks
152 of the target and are composed of gigabreccia (fragments size from 10–50 m up to 300–500 m),
153 megabreccia (1–10 m) and block breccia (0.2–1.0 m) with wide distribution of shatter cones. The
154 entire unit is 0.7–0.9 km thick at the crater center.

155 *Suevites* occur as a 0.8–1.0 km thick unit that covers the allogenic breccias and locally the
156 Paleozoic sediments. The size of the breccia fragments increases toward the bottom of the unit,
157 they consist mostly of litho-clasts (up to 85% among the fragments). Suevites occur as lappile-
158 agglomerated, agglomerated, and rarely as block vitro-clastic types. The latter is characterized by
159 15–30 vol% of “glass” with isometric, elongated and irregular shapes from small drops up to
160 meter-sized bodies with spot-like, striped and fluidal texture. Litho-clasts within the “glasses”

161 can be fully melted or just preserved as blurred objects. Lapille and lapille-agglomerated litho-
162 vitro-clastic sintered suevites are very rare, and consist of unsorted clastic material with 50–80%
163 “glass” and 20–50% litho-clasts.

164 *Tagamites* are locally spread, mostly in the basin of the Anaroga river, forming lava-like
165 bodies; they occur as lens and layer-like bodies with complicated shapes up to 15 m thick with
166 up to 200–300 m horizontal extension. They consist of “glasses” with a total loss of the host
167 sediment texture. Tagamites have essentially a lower content of clastic rock- or mineral
168 fragments, ranging from 5 to 35 vol% of the whole rock. They have lens-spot, spot-fluidal
169 textures, resulting from different levels of crystallization.

170

171 **2.4. Target carbonaceous matter characterization**

172 According to Yudovich et al. (1990), the carbonaceous matter of the Pay-Khoy is mostly
173 either finely dispersed in black shales or occurs in concentrated form in coal layers with a
174 thickness of about 10 cm. Besides, our own field observations revealed that the coal layers
175 within the target area could be at least up to 2 m thick (Fig. 2d). This finding is based on the
176 presence of quite large coal fragments within suevites that are exposed at the surface, where they
177 are intensely weathered and disaggregated.

178 In coeval black shales from the southern part of the Pay-Khoy Ridge, containing from 1 up
179 to several wt% C_{org} , Yudovich et al. (1990) report low bitumen content and high concentrations
180 of aromatic rings in the carbonaceous matter. The largest C concentration was measured at about
181 30 wt% for Late Devonian clay-siliceous shales. Transmission electron microscopy (TEM) data
182 of the carbonaceous matter revealed globular-like aggregates made of small globules 6 nm in
183 diameter, with the diffraction patterns having 3.37 Å d -spacings and wide rings with 2.02 and
184 1.17 Å d -spacings. The exothermal oxidation effect (in differential thermo-gravimetric analysis,
185 DTG) for the carbonaceous matter started between 500–600°C with double maxima in some
186 cases. Thus, Yudovich et al. (1990) have described the matter as comprised of two differently

187 ordered carbon types – an amorphous/shungite-like carbon and a turbostratic graphite carbon.
188 The first type was assigned a bathyal sedimentary origin, having an autochthonous nature,
189 whereas the graphitic matter (turbostratic graphite) was considered to be para-autochthonous,
190 resulting from a migration process.

191 **2.5. Previous work on Kara impact carbons**

192 In previous studies (Yezerkiy, 1982, 1986; Reshetnyak & Yezerkiy, 1990), after-coal impact
193 carbons were studied only for size fractions larger than 0.5 mm. Yezerkiy (1986) described
194 several varieties of after-coal carbons chemically extracted from Kara astrobleme impactites. He
195 described them as irregularly shaped fragments with sizes up to 1.6 mm, sometimes rounded
196 grains, many of which had an optically visible friable and porous structure. Some grains had a
197 hard covering but were empty inside. Some carbon particles occurred as optically transparent
198 amber-like fragments, while others were opaque with black, gray, brownish, light blue or white
199 color. Among them, the dense particles were harder than moissanite, and the porous varieties
200 could be easily broken. Three types of carbonaceous grains were described on the basis of color
201 – A, B and C (Table 1), which occur as independent particles or as aggregates consisting of
202 different varieties and with gradual transitions between them. The phase-state determination of
203 the individual carbons was made on the basis of Laue X-ray studies.

204 Variety A, with black color, was described by V.A. Yezerkiy as an intermediate state
205 between coal and diamond. Varieties B and C were identified as diamonds, where B occurred as
206 brown grains (subvariety B₁) and aggregates of brownish, white and blue layers (subvariety B₂).
207 Variety C was categorized as white grains, with a texture recognized by optical observations and
208 Laue X-ray studies. This variety might occasionally contain lonsdaleite (Yezerkiy, 1986).

209

210 In the following, on the basis of high-resolution techniques, we describe the most
211 important carbon phases (Table 1, Figs. 3–9) that are reliably characterized for the impact-rock
212 fractions with sizes <2 mm, including the diamond size fraction <0.5 mm. Attention is paid to

213 the description of the after-coal diamonds. In addition, other carbon phases that are syn-genetic
214 to diamonds within the studied impact rocks, including abundant glass-like carbon and rare
215 impact graphite, are described here briefly but are subjects for future, more detailed studies. Our
216 classification of Kara impact carbons is based on structural state and morphology; the
217 correspondence with earlier categorizations (Yezerkiy, 1986) and some general characteristics
218 of carbons are given in Table 1.

219

220

221 **3. Material and methods**

222 *Studied material.* The material was sampled in 2015 by the Russian team during a field
223 expedition to the southern part of the Kara astrobleme, covering the region of the Anaroga, Kara,
224 Sopcha-Yu and Sibirchata-Yacha river basins. We focus here on material from the regions of the
225 Anaroga and Kara rivers. The study combines preliminary standard mineralogical, petrological,
226 and geochemical investigations with a detailed mineralogical study of carbon phases using high-
227 resolution methods for separated particles and *in situ* inclusions in rocks.

228 For *in situ* studies we have prepared fresh mechanically crushed surfaces and polished thin
229 sections for studies in transmitted and reflected light and for instrumental measurements. Non-
230 diamond-containing materials such as Cr_2O_3 and Sr_2O were used for polishing. Preliminary
231 epoxy impregnation of the suevites was prepared, without the use of any covering glass and
232 balsam covering of the thin sections. For the studies of the carbonaceous matter many
233 measurements were carried out with control of fresh crushed rock surfaces and extracted grains
234 in order to avoid the possibility of contamination by epoxy glue.

235 The carbon phases were extracted from the host impact rocks at the Laboratory of
236 Diamond Mineralogy at the Institute of Geology of Komi Scientific Center of Russian Academy
237 of Sciences (IG Komi SC UB RAS, Syktyvkar, Russia) using the thermochemical dissolution
238 technique used for microdiamond concentration. The technology allows enriching

239 microdiamonds with sizes $>1 \mu\text{m}$; it includes a complex multiple-stage chemical treatment such
240 as boiling in a mixture of concentrated H_2SO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$ aqueous solution, melting with
241 NaOH at 500°C , treatment by HCl aqueous solution. We used small-volume dissolution with an
242 initial rock mass of 5 g per sample. The concentrated particles were picked up from the filters
243 under an optical binocular microscope MBS-10 at magnification $\times 40$, and then were studied in
244 detail by the methods described below.

245 *Optical observations* were made using Polam 312 and Olympus BX41 microscopes in
246 transmitted and reflected plane- and cross-polarized light with objectives $\times 4.7$ – 100 (IG Komi SC
247 UB RAS, Syktyvkar, Russia).

248 *Raman spectroscopy* analysis was performed at IG Komi SC UB RAS using a high-
249 resolution Raman spectrometer LabRam HR800 (Horiba Jobin Yvon, France). Spectra were
250 collected *in situ* from surfaces of fresh crushed impact rocks, polished thin sections, and from
251 individual chemically concentrated grains. An Ar^+ excitation laser with 488-nm wavelength was
252 used to collect spectra in the range 100 – 8000 cm^{-1} at room temperature, with 1.2 mW laser
253 power. The use of $\times 50$ and $\times 100$ objectives and a grating with 1800 grooves/mm yielded
254 approximately $1\text{-}\mu\text{m}$ spatial and 1-cm^{-1} spectral resolutions.

255 An experimental treatment aimed at recovering diamond from grains coated by glass-like
256 carbon was applied by *in situ* laser combustion of the coating. In such case the laser power was
257 raised until the outer glass-like cover was burnt and the diamond core appeared. Usually the final
258 laser power output for combustion experiments was no more than 12 mW, but reached 72 mW in
259 some cases. The diamond phase was confirmed by Raman spectra (T_{2g} scattering mode) and
260 optical observations.

261 To ascertain the diamond nature of the after-coal diamonds, ultraviolet (UV) Raman
262 spectroscopy studies were performed at the SSC FSUE Keldysh Research Centre (Moscow,
263 Russia) with a T64000 (Horiba Jobin Yvon, Japan) Raman spectrometer using a 244 nm laser
264 excitation wavelength with a grating of 2400 grooves/mm, and a $40\times$ objective. To prevent laser-

265 induced sample damage, the laser power on the sample was decreased to ~5 mW power, while a
266 laser spot diameter ~5 μm and an exposure time of 5 min was used.

267 All spectra were checked by pre- and post-observations with an optical microscope to
268 check for possible specimen changes after laser exposure. After background correction, all
269 spectra were deconvoluted to individual peaks using a curve fitting procedure by Gaussian and
270 Lorentzian functions from the software provided by LabSpec 5.36.

271 *Scanning electron microscopy and microprobe analysis (EMP).* Analysis of the impact
272 rocks and carbonaceous matter was made at the IG Komi SC UB RAS with *in situ* observations
273 in fresh rock surfaces and carbon-grain concentrates using a VEGA 3 TESCAN scanning
274 electron microscope (Tescan, Czech Republic) with a VEGA 3LMN, INCA ENERGY 450
275 energy-dispersive detector that was used for mineral chemical and morphological analysis. The
276 measurements of elemental contents were done without any conductive covering to avoid
277 analysis contamination when analyzing carbon *in situ*. Afterward, the diamond specimens were
278 covered by a conductive thin carbon film for high-quality SEM images.

279 *Atomic force microscopy.* An atomic force microscope Integra Prima (NT-MDT, Russia)
280 was utilized to study fresh surface morphology. Silicon cantilevers with 25 nm-thick conducting
281 Pt coating and 20 nm tip radius (PPP-CONTPt, Nanoworld) were employed at room temperature
282 (25–27 °C) and about 65% relative humidity. The diamond grains were set on electrical
283 conducting glue without any conducting covering. The average nanocrystallite sizes were
284 calculated with statistical analysis of the morphological nanoimages.

285 *Transmission electron microscopy.* The preliminary TEM studies were conducted at the IG
286 Komi SC UB RAS with powder specimens of the carbon particles concentrates using a Tesla BS
287 500 (Czech Republic) TEM at 60 kV accelerating voltage. Specimens were prepared by
288 ultrasonic suspension in ethanol and crushing of individual diamond grains with settling on a
289 holey carbon supporting film.

290

291 **4. Description of the host-rocks**

292 Here we briefly describe the features of the Kara impact rocks – host tagamites and
293 suevites containing after-coal diamonds and associated carbons – sampled by our team in the
294 Anaroga and Kara regions.

295 **4.1. Tagamite**

296 Tagamite was sampled at the Anagoga River from a large layer-like body with a maximum
297 visible thickness of 7–8 m (Fig. 2a), with occasional inclusions of large coal fragments at the
298 bottom of the body (Fig. 2b) and microsized fragments within the melt rock matrix (Fig. 2a,
299 inset). Tagamite has light brown color, massive texture with fluidal rock structure mostly
300 comprised by “glass” - solidified partly crystallized impact melt (up to 80–90 vol%) with rare
301 relics of the target rocks such as sandstones, minerals and coals (Fig. 2a, b, insets).

302 The rock clasts in tagamite have rounded edges, but the shape of the initial clast may be
303 recognized as a melt-treated outer zone. The relict clasts are generally pieces of sandstones with
304 sizes up to 8–10 cm, and rarely shales or minerals. The tagamite “glass” is essentially slightly
305 altered due to the ~70 m.y. elapsed since the impact event. The “glass” appears opaque, as a
306 crypto-crystalline gray-brown, quite soft matter, which easily disaggregates during drying at
307 room conditions. It contains less than 1 wt% of carbonaceous matter, according to DTG analysis.
308 The “glass” generally consists of a cryptocrystalline alumino-silicate substance represented
309 mainly by albite according to X-ray diffraction and Raman spectroscopy. The specific feature of
310 the tagamite matter is the essential presence of molecular water in a chemically unbound state. A
311 true glass, either in tagamites or in suevites, is characterized generally by aluminosilica and SiO₂
312 and was rarely found, mainly as thin rims surrounding pores within the rock. According to our
313 field observations and petrochemical studies, we assume that tagamites were formed after target
314 sandstones, consistent with the earlier studies of Selivanovskaya et al. (1990).

315

316 **4.2. Suevites**

317 4.2.1. *The suevites syngenetic to tagamites*

318 The suevites syngenetic to tagamites (Fig. 2a) are represented by layer-like lens bodies
319 with a visible maximum thickness of 6–7 m. They have a macroscopically obvious breccia-like
320 structure with variable content of the target fragments consisting mainly of sandstones and
321 siltstones, making up to 50-60 vol% of the total rock. The suevites have brownish color similar
322 to the tagamites and show a transition to tagamites. The “glass” is well recognized within the
323 suevites by their fluidal texture (Fig. 2a, inset), it preserves some relics of coal fragments. After
324 dissolution we have found that a 5-g standard sample of tagamite and connected suevite includes
325 at least 5–10 diamond grains with sizes up to around 50–100 μm .

326 4.2.2. *The suevites unconnected to tagamites*

327 Suevite sampled on the right bank of the Kara River is essentially different from that
328 described above. It has gray color and an essentially porous structure with a lot of micro- and
329 macrogeodes filled by calcite and quartz. The suevite matrix consists predominantly of
330 carbonate, and the breccia fragments are comprised of black shales, limestones and coal
331 fragments (Fig. 2c). The coal fragments can reach 2 m in size (Fig. 2 d). At the same time some
332 quite large sandstone fragments and light-colored “glass” occur within the suevite, reaching
333 several decimeters in thickness and >1 m in length.

334 According to our field observations and microscopic studies, we assume that the Kara
335 suevite was formed from carbonate sediments, black shales and some part of coal layers of the
336 target. For a detailed study, part of a “glass” lens-like body about 1.5 m long and 15 cm thick
337 was dissolved for diamond recovery. As a result it was estimated that every 5 g of the “glass”
338 included at least five fully crystalline diamond grains up to 100–200 μm in size, as well as up to
339 200–300 diamonds in the form of after-organic paramorphs with a similar particle-size range. On
340 the basis of the number of extracted grains, the after-coal diamond concentrations can be
341 expressed to a standard diamond unit: up to 85 carat per ton for the impact rock, and up to 3000
342 carat per ton in the solidified impact melt.

343

344 **5. Results: impact carbons**

345

346 **5.1. Diamonds**

347 Kara after-coal diamonds extracted from chemical concentrates occur as polycrystalline
348 grains up to 200 μm in size (Fig. 3–5). On the basis of morphological features we have divided
349 them into two main varieties of after-coal diamonds – micrograined (sugar-like, subdivided into
350 coherent and friable) and paramorphs after organic relics. Here we describe the mineralogical
351 features of the after-coal diamonds; their technological properties will be addressed elsewhere.

352

353 *5.1.1. “Sugar-like” diamonds*

354 The diamonds occur mainly in irregular shapes, mostly with rounded edges up to wholly
355 rounded forms (Fig. 4a-c). Rarely they have a layered texture and morphology preserved after
356 organic relics (Fig. 4f). This diamond type occurs at about 85 carats per ton in the studied impact
357 rocks. Although they are usually white in color, they can also have gray and slightly gray,
358 sometimes gray-blue and brownish shades. They generally have optically recognizable
359 micrograined “sugar-like” surfaces without optically noticeable “diamond” luster, and only
360 rarely have smooth surfaces with a medium greasy luster. We call these diamonds “coherent
361 sugar-like diamond”.

362 Raman studies with visible laser excitation (Fig. 6, left column) over a wide spectral range
363 show that the “coherent sugary” diamonds are characterized by numerous luminescence bands
364 on the basis of high total luminescence intensity. In Raman spectra obtained with UV laser
365 excitation the diamond band corresponding to T_{2g} stretching is at the standard position of 1332
366 cm^{-1} , with a full-width at half maximum (FWHM) of about 10–30 cm^{-1} (Fig. 7). In spectra taken
367 using visible laser excitation, a band around 1600 cm^{-1} belongs to a G-band of sp^2 -carbon; the
368 features at 1450-1475 and 1710 cm^{-1} are observed as luminescence bands.

369 Among the light-colored diamond varieties with different shades of color, there are
370 numerous “friable sugar-like diamonds” that occur as opaque, irregularly shaped polycrystalline
371 grains without rounded edges grains nor bright luster. With visible laser excitation they have an
372 unrecognizable diamond Raman band concealed by an intense luminescence (Fig. 6, spectra 2–
373 4). The UV Raman spectra demonstrate that the white friable opaque carbon micrograins are
374 indeed polycrystalline diamonds (Fig. 7).

375

376 *5.1.2. After-organics diamond paramorphs*

377 In our study brown diamonds are characterized as optically transparent matter having
378 surfaces with bright lustre that occur mostly as paramorphs after organics, preserving relict
379 morphology of the initial organic matter as shown by optical microscopy or SEM (Fig. 3 b-d, 4 i-
380 l). They occur in tagamite and suevite samples, in impact “glass” and are accompanied by impact
381 “sugar-like” diamond and glass-like carbon. In impact “glass” the brown diamonds are very
382 abundant, comprising up to a half of the carbon particles within a chemical concentrate, yielding
383 up to 200–300 particles per 5 g sample of a “glass” from suevite (up to several thousand carat
384 per ton, as noted above).

385 In transmitted cross-polarized light the diamond is anisotropic without optical extinction.
386 Due to a very high luminescence background, the diamond Raman band at 1332 cm^{-1} is usually
387 concealed when excited by the visible laser (Fig. 6, spectra 3, 4), which does not allow a
388 standard phase-state analysis. However, in some cases the diamond line can be recognized in the
389 brown carbon despite the high luminescence (Fig. 6, spectrum 4). For all studied cases the
390 comparison of white and brown diamond luminescence allowed observing similar band features.

391 The UV Raman measurements were the most useful for phase determinations, and clearly
392 demonstrated the diamond nature of the after-organic diamond paramorphs (Fig. 6, right
393 column). However, this diamond type shows a large downshift of the diamond band ($1317\text{--}1326$
394 cm^{-1}) relative to the standard value of 1332 cm^{-1} (Fig. 7), and also shows a wide associated

395 FWHM of about 40–60 cm^{-1} , which is usual in nanocrystalline diamond (Yoshikawa et al., 1993,
396 1995; Prawer, 2000; Ferrari, 2004; Piscanec et al., 2005; Vul', 2006; May et al., 2008; Osswald
397 et al., 2009). The downshift due to heating is larger for nanosized material than for regular-sized
398 one (May et al., 2008; Isaenko & Shumilova, 2008).

399 Debye-Scherrer X-ray analysis supports the identification of the brown after-organics
400 paramorphs as diamond. The X-ray pattern shows a single small-intensity ring, without any hint
401 of spots at all, corresponding to polycrystalline diamond with a spacing $d_{111} = 2.05 \text{ \AA}$. Electron
402 diffraction patterns of the brown diamonds are characterized by three rings corresponding to
403 (111), (220) and (311) planes of cubic diamond (Fig. 5a). No texture was recognized by electron
404 diffraction. Preliminary tests of the mechanical properties show high fragility of after-organics
405 diamond paramorphs and very high hardness of their crushed fragments.

406

407 *5.1.3. Kara diamonds nanostructure*

408 Here we describe the results of our comparative study of white coherent sugar-like and
409 brown after-organics diamonds (Fig. 8), which were selected for a detailed analysis by atomic
410 force microscopy (AFM) as they have essentially different morphological and spectroscopic
411 features. According to AFM observations all impact after-coal diamonds occur as
412 nanopolycrystalline aggregates with different degrees of crystallinity, varieties of shapes, and
413 sizes of nanocrystals (Fig. 8).

414 The white “sugary” diamonds (Fig. 8 a, b) are characterized by aggregates of 20–40 nm
415 nanocrystallites tightly joined together into larger units reaching 80–100 nm in size, or forming
416 continuous aggregates.

417 The after-organics diamond paramorphs are quite diverse in their crystallinity (Fig. 8 c–f).
418 In a first case they are characterized by poorly distinguishable nanocrystallites, from a few tens
419 of nanometers up to 150 nm in size (Fig. 8 e, f). Another case of after-organics diamond contains

420 small octahedral nanocrystallites with an average size about 140 nm that are distributed more
421 uniformly through the nanopolycrystalline aggregate (Fig. 8 c, d).

422 All the recognized Kara-impact after-coal diamond types are characterized by quite a
423 rough surface at the nano-scale, with deeply porous structures and an almost complete absence of
424 carbon cement between the crystallites. The AFM results are consistent with TEM observations
425 and electron-diffraction data in showing that the Kara after-coal diamonds occur as
426 polycrystalline diamond grains devoid of any layered texture in the aggregates (Fig. 5a).

427

428 **5.2. Graphite**

429 Novel, quite rare graphite particles formed under impact process have been recognized in
430 suevites connected to tagamites. The graphite occurs as polycrystalline aggregates (Fig. 3 e, f)
431 with sizes up to several hundred micrometers, or as individual micro- and nano-particles
432 revealed by TEM and Raman spectroscopy. The particles usually have irregular shapes with a
433 flattened morphology for individual grains. The polycrystalline nature of graphite particles has
434 been established by optical microscopy (Fig. 3 e, f), electron diffraction (Fig. 5b) and Raman
435 spectroscopy (Fig. 6, spectrum 6) on the basis of relative intensity of the D band and FWHM of
436 the G band of sp^2 carbon, which are inversely proportional to the graphite crystallite size
437 (Tuinstra & Koenig, 1970; Wopenka & Pasteris, 1993; Ferrari & Robertson, 2004). The first-
438 order region of the Raman graphite spectrum (Fig. 6, spectrum 6) consists of a strong and narrow
439 1581 cm^{-1} line of the E_{2g} graphite mode with a FWHM of 26 cm^{-1} which corresponds to ~ 10 –
440 30 nm crystallite sizes. The 1358 cm^{-1} band belongs to a disordered feature of graphitic carbon,
441 named the D-band. The second graphite order is present in the spectrum with the 2441, 2741,
442 2949 and 3245 cm^{-1} bands.

443

444 **5.3. Glass-like carbon**

445 Glass-like carbon (GLC) is the most abundant impact carbon phase. It occurs in black
446 grains with sizes up to several hundred micrometers with irregular shapes as breccia-like micro-
447 fragments. They have glassy surfaces with bright reflection and, much more rarely, occur as
448 paramorphs after organic matter similar to the described diamond grains (Fig. 4c, 4 i-l). The
449 phase state of the glassy carbon variety was characterized by high-resolution Raman
450 spectroscopy (Fig. 6, spectrum 5). The bands at 1359 and 1598 cm^{-1} correspond to D and G
451 bands of sp^2 carbon, the bands in the region 2400–3300 cm^{-1} belong to a second order of the
452 graphitic carbon. These spectroscopic features are in a good correspondence with industrially
453 produced glass-like carbon (Harris, 1997).

454

455 **5.4. After-coal diamonds with GLC coats**

456 Among products recovered by chemical dissolution, special attention was devoted to
457 diamonds coated by glass-like carbon. They look like usual GLC fragments with irregular
458 morphology and bright reflectance, yet at the same time they are denser than usual GLC. During
459 detailed *in situ* Raman studies the diamonds were exposed by intense laser heating directly under
460 the microscope, as explained in the Procedure section. Thus, after burning the GLC covering, a
461 diamond core could be exposed and became visible by optical microscopy, with the
462 corresponding appearance of the Raman diamond band at 1318-1321 cm^{-1} (Fig. 9), downshifted
463 due to overheating by the laser heavy treatment (May et al., 2008; Isaenko & Shumilova, 2011).
464 The same results were produced by several similar experiments to prove the burning of the GLC
465 coating and exclude GLC-to-diamond transition under laser treatment.

466

467

468 **6. Discussion**

469 **6.1. The status of “togorite”**

470 Based on the observations of this study we have distinguished among the proven after-coal
471 impact carbons: glass-like carbon, diamond (two varieties with two subvarieties), and graphite
472 (Table 1). The typomorphic features of diamonds coated by GLC demonstrate that they are
473 similar to the previously described “togorite” (Yezerkiy, 1982; variety A in Yezerkiy, 1986),
474 proposed as a new carbon mineral (non approved by the IMA). The aggregates comprised of
475 diamond (core) and glass-like carbon (coat) should provide macroscopically high hardness and
476 density in accord with the relative proportions of diamond and GLC components. We therefore
477 conclude that the intermediate state between diamond and coal named “togorite” by Yezerkiy
478 (1982) is just diamond coated by GLC, and so does not represent an independent carbon phase.

479

480 **6.2. After-coal diamond features, after-organic diamond paramorphs**

481 Among the recognized after-coal diamonds, we have categorized two varieties that
482 essentially differ from after-graphite impact diamonds: (i) *sugar-like* fully crystalline-ordered
483 nanocrystalline aggregates *with coherent and friable subvarieties*; (ii) *after-organic paramorphs*,
484 consisting in nanocrystalline–sub-nanocrystalline–ta-C aggregates with various proportions of
485 crystalline and amorphous carbon. The varieties are well distinguished between them by a
486 complex data and differ from the earlier published data on after-coal diamonds (Yezerkiy, 1986;
487 Reshetnyak & Yezerkiy, 1990) (Table 1, Fig. 7).

488 The after-organics diamond paramorphs have been found for the first time. These grains
489 have specific typomorphic features that differentiate them from the other after-coal diamonds –
490 morphology, optical transparency, brown color, very high luminescence, spectral features,
491 nanocrystalline state, ordering level. Considering the well-preserved morphology of organics one
492 may assume that the diamond paramorphs were formed directly from organics. However, our
493 rare finds (only two grains so far) with clearly recognizable relics of framboidal pyrite (Fig. 4 j,
494 k) suggest instead that at least part of the post-organics diamond paramorphs were formed after

495 partially destroyed organics, perhaps in the form of lignin, which is able to preserve relics of the
496 initial organics morphology within sedimentary rocks.

497 Additionally we have to discuss about the poorly ordered diamond and carbon matrix
498 within the after-organic diamond paramorphs. According to the UV Raman spectra (Fig. 6,
499 spectra 3, 4; Fig. 7) the diamond band 1317–1326 cm^{-1} is observed with a significant downshift
500 from the standard T_{2g} mode of cubic diamond. Many scientists ascribe such a shift to lonsdaleite
501 (*e.g.* Smith & Godard, 2009, 2013; Karczewska, 2010; Goryainov et al., 2014; Jones et al.,
502 2016). However, the downshift can be explained in first place by the nanocrystalline nature of
503 the impact diamonds (Ferrari, 2004; May et al., 2008; Osswald et al., 2009; Vul`, 2006;
504 Yoshikawa et al., 1993, 1995). In many studies explaining the downshift by the presence of
505 lonsdaleite (*e.g.* Smith & Godard, 2009, 2013), the used laser excitation was quite high, which
506 has to result in essential overheating (May et al., 2008; Isaenko & Shumilova, 2009; Shumilova
507 et al., 2014), with a large red-shift of the diamond line. At the same time, the nanocrystalline
508 nature of the impact diamond studied by Smith & Godard (2009), as shown by the large FWHM
509 ($54\text{--}96 \text{ cm}^{-1}$) in the Raman spectra, can provide a diamond-line downshift explained by a phonon
510 confinement effect (Yoshikawa et al., 1993, 1995; Osswald et al., 2009). Thus, considering the
511 various possible reasons for the Raman diamond-line downshift, the presence of lonsdaleite
512 cannot be established on this simple basis but requires independent additional high-resolution
513 studies. In many published cases the lonsdaleite presence is not evidently proven.

514 In our case, on the basis of X-ray diffraction, electron diffraction and AFM observations,
515 we did not recognize either lonsdaleite or any texture like in after-graphite impact diamonds
516 (Kvasnytsya & Wirth, 2013; Masaitis et al., 1998). Thus, we conclude that the large Raman
517 diamond-line downshift in after-organic diamond paramorphs is due to their nanocrystalline
518 structure. Yezersky (1986) mentioned rare particles with texture and the very rare presence of
519 lonsdaleite among after-coal diamonds, but without detailed description, so that it is not possible
520 to ascertain their correctness.

521 In the UV Raman spectra of after-organic diamond paramorphs (Fig. 6, spectra 3, 4), the
522 wide band centered at 1225–1260 cm^{-1} is assigned as a red-side shoulder of the certainly
523 narrower diamond band at 1317–1326 cm^{-1} . This shoulder cannot be a D band of graphitic
524 carbon as it must not be present for the used UV laser excitation (Ferrari, 2004). According to
525 previous studies (Yoshikawa et al., 1993, 1995; Prawer, 2000; Osswald et al., 2003; Ferrari,
526 2004; Piskanec et al., 2005; Vul', 2006), this shoulder band belongs to nanocrystalline diamond
527 with crystallite sizes of about several nanometers (sub-nanocrystalline diamond) and smaller –
528 up to amorphous tetrahedral diamond-like carbon (ta-C) with some presence of amorphous sp^2
529 carbon at the nanocrystallite boundaries, as indicated in the UV Raman spectra by a wide G band
530 (1600–1650 cm^{-1}). The sub-nanocrystalline diamond, ta-C and amorphous sp^2 carbon, in some
531 cases, are probably accompanied by a C=O band (about 1730 cm^{-1}).

532 It is interesting that similar Raman features have been described for meteoritic
533 nanodiamonds from Efremovka (CV3) and Orgueil (CI) chondrites (Shiryaev et al., 2011) and
534 synthetic detonated nanodiamonds (Vul', 2006).

535 **6.3. Proposed mechanism of after-coal diamond formation**

536 Analysis of after-coal diamonds features clearly shows essential difference from after-
537 graphitic impact diamonds: specific morphology of rounded grains and post-organic paramorphs;
538 porous micro- and nano-structure; the absence of texture and the presence of well-formed
539 octahedral diamond nanocrystals of the polycrystalline grains.

540 Based on the features of the initial carbonaceous matter and the established typomorphic
541 features of after-coal diamond, we propose a formation mechanism alternative to the well-known
542 direct graphite-diamond transition (diffusion-less) (*e.g.* Lonsdale, 1971; Kurdumov et al., 2012;
543 Xie et al., 2014) and to the poorly known diffusion-limited reconstructive mechanism
544 (Borimchuck et al., 1991).

545 According to Yezerskiy (1986) the initial coals of the Kara target include 1.42–2.73 % of
546 hydrogen and 0.8–1.49 % of nitrogen (oxygen content was not measured, but can reach several

547 percent for the relevant coal stages – from bituminous to anthracite). Thus, diamond formation
548 from the coal precursor implies first of all the carbonization of the coal matter. The few studies
549 available on transformations of coal and organic compounds under shock pressure (Tyburezy et
550 al., 1986; Borimchuck et al., 1991; Peterson et al., 1997; Kurdyumov et al., 2009; Korochantsev,
551 2012) allow one to predict possible transformations. According to shock high temperatures of
552 large impacts (*e.g.* Masaitis et al., 1998; Langenhorst, 2002) and published experiments
553 (Korochantsev, 2012), we suppose a high-temperature pyrolysis as the initial stage of coal
554 transformation, with destruction of long-range organic compounds (preferably of aliphatic
555 varieties), resulting in fluid phase removal (as supported by observed pores within diamond
556 aggregates) and formation of amorphous pure carbon and soot (Borimchuk et al., 1991;
557 Kurdyumov et al., 2009). The high shock temperatures up to 4000 K (Schmitt et al., 2005;
558 Kurdyumov et al., 2012, 2013) can allow some short-distance diffusion of amorphous carbon,
559 similar to experimentally produced process (Borimchuk et al., 1997; Kurdyumov et al., 2013)
560 with fast diamond/graphite crystallization to idiomorphic lonsdaleite-free/texture-free
561 nanocrystallites and carbon polymerization at quenching to glass-like carbon, as analysed in the
562 impact products by Raman spectroscopy and TEM.

563 **6.4. Implications: relation to carbonado**

564 According to our data and Yezerskiy's study (1986), the after-coal diamonds from the Kara
565 astrobleme are quite similar to carbonado, including a light isotopic carbon composition, a
566 lonsdaleite-free, texture-free and porous structure, lack of crystallites of preferred orientation,
567 presence of well-shaped crystallites, and blue luminescence. After-coal diamonds are therefore a
568 strong piece of evidence in support of the impact hypothesis for the origin of carbonado, whose
569 nature has been debated for decades (*e.g.* Kaminsky, 1979; 1987, 1991; Smith & Dawson, 1985;
570 Yezerskiy, 1986; DeCarli 1987; Galimov et al., 1987; Heaney et al., 2005; Kagi et al., 2007;
571 Rondeau et al., 2008; Petrovsky et al., 2010; Piazzolo et al., 2016). Some difference in mineral
572 inclusions and elemental impurities in known carbonados could be a result of different target in

573 different parts of the Earth's crust. A detailed comparison between after-coal diamond and
574 carbonado is scope for future work.

575

576 **7. CONCLUSION**

577 On the basis of the provided complex study we have corrected understanding of carbon
578 phases within Kara astrobleme impact products presented by glass-like carbon, graphite and 2
579 types of after-coal diamonds – sugar-like (with 2 subvarieties) and after-organics paramorphs.
580 The tight growths of diamond with glass-like carbon are quite often, where the latter forms
581 covering of a diamond core.

582 The novel afetr-organic diamond paramorphs have been found for the first time and
583 described in detail. This type of impact diamond is widely spread through the Kara impact rocks
584 locally bringing the huge concentrations up to first thousand carat per ton.

585 Our results demonstrate for the first time the nanostructural features of after-coal impact
586 diamonds describing their enigmatic character, which is quite different from well-studied and
587 widely known after-graphitic impact diamonds by idiomorphic nanocrystallites within diamond
588 aggregates, presence of rounded shapes and after-organics morphology with nice preserved
589 nanorelief, texture and lonsdaleite absence, porous micro- and nanostructure, presence of some
590 sub-nanocrystalline–ta-C carbon matrix. The lack of lonsdaleite is one of the most important
591 typomorphic feature of the after-coal impact diamonds, which in a complex of the other listed
592 features makes them similar to carbonado. Thus, future detail comparison of after-coal impact
593 diamonds and carbonado can help understand the latter secret.

594 The observations support probability of at least two stages of diamond formation through
595 pyrolysis and subsequent diffusion-limited reconstructive mechanism of carbon.

596 The Kara impact object can be a basis for fundamental studies of after-coal impact
597 diamond which may even be more widely spread through the Earth than after-graphitic ones.
598 Thus, the presented data can be used for searching for new prospective astroblemes with huge

599 diamond resources at large impact craters around the world where some coal, peats or even just
600 living organics are present in the target matter.

601

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613

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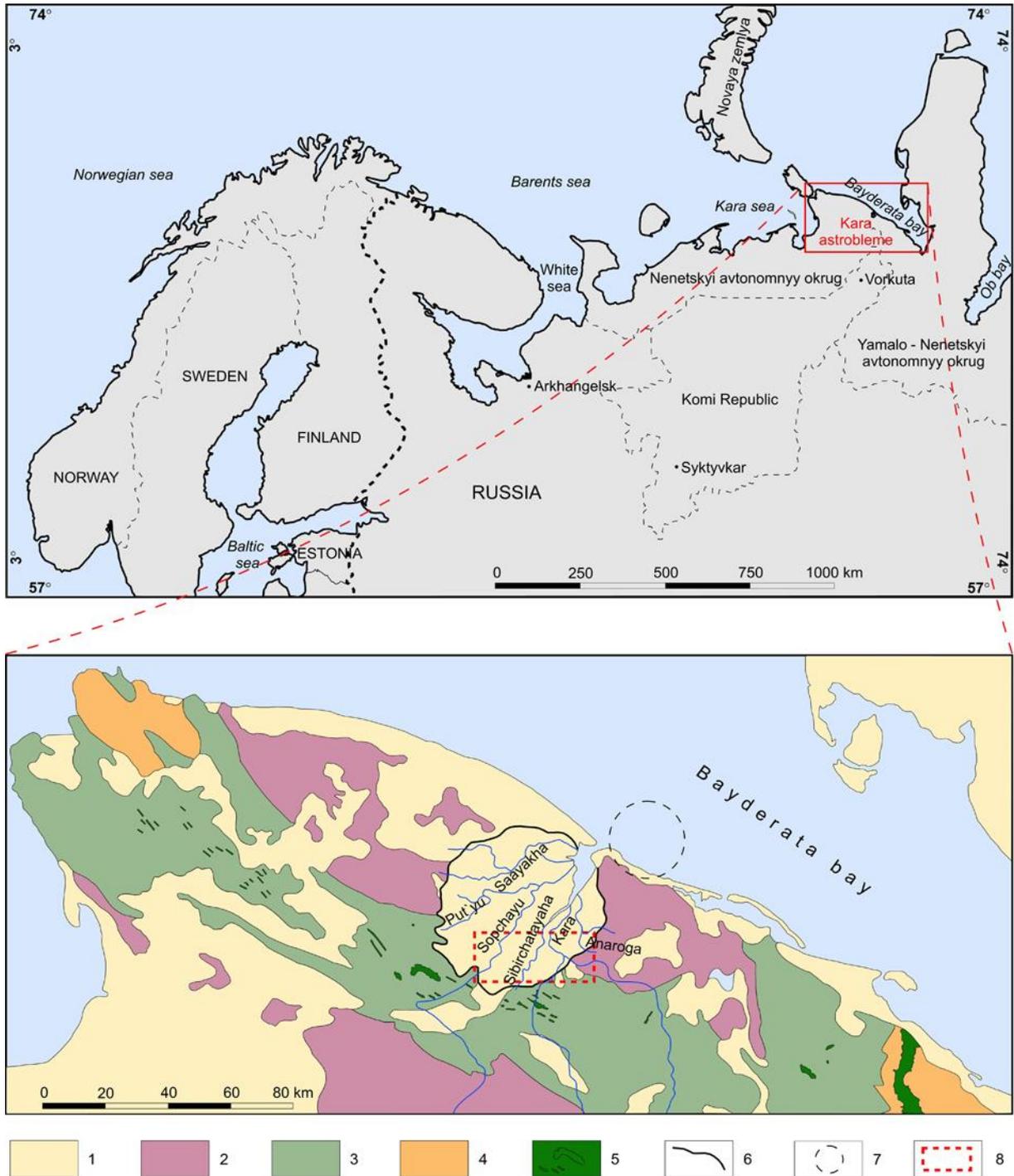
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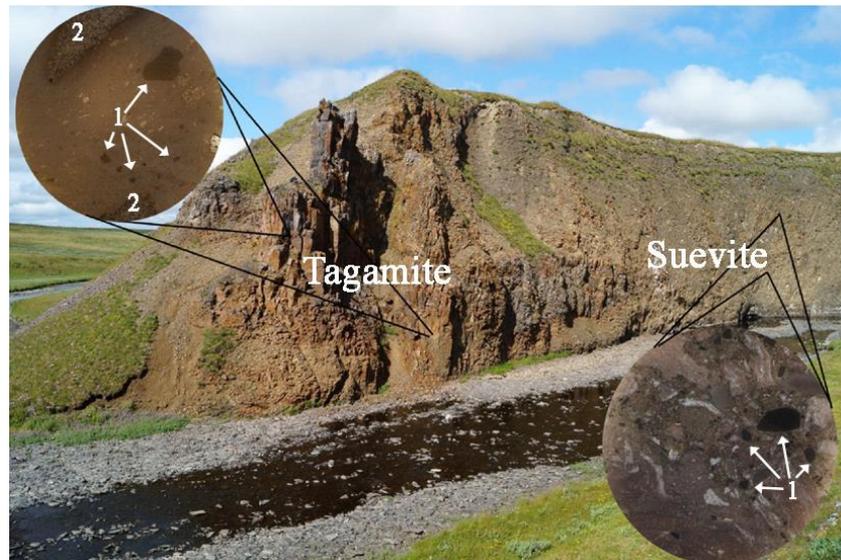
Figures



806 Fig. 1. Geological scheme of the Kara region territory modified after Yudovich et al. (1998) and
807 Mashchak (1990) with additions: 1 - Pliocene-Quaternary sediments; 2 - Triassic and Permian
808 deposits; 3 - Carboniferous, Devonian, Silurian and Ordovician sediments; 4 - Upper Proterozoic
809 sediments; 5 - Late Devonian tabular body and dikes of dolerite and gabbro-dolerite; 6 - the
810 border of the Kara astrobleme; 7 - the border of the Ust-Kara astrobleme; 8 - sampling region.
811 Upset – geographic setting of the studied area. The color image is available within an on-line
812 version.
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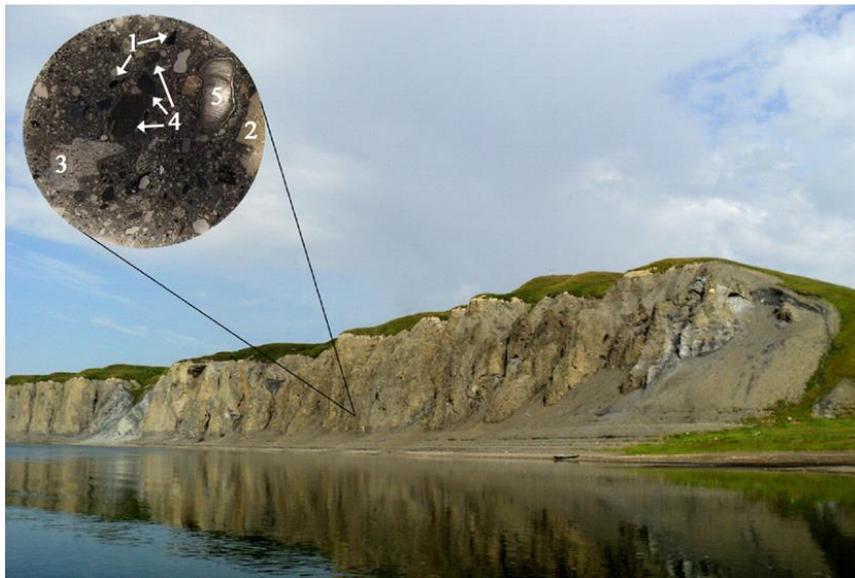
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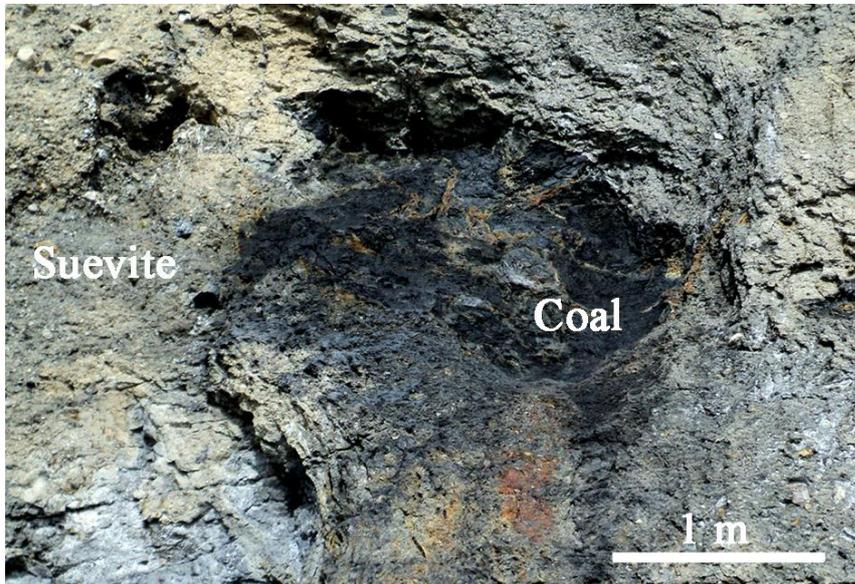
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825 Fig. 2. Impactites of Kara astrobleme. a – tagamite (left inset) and co-following suevite
 826 after silicate rocks (right inset) from the outcrop maximum high of 20 m at the Anaroga River, b
 827 – 1x2 m coal fragment (dark grey, in a center) within the bottom part of the tagamite body, the
 828 Anaroga River. c – Suevite after organic-carbon-bearing carbonates and black shales, the visible
 829 outcrop length is about 700 m, at the Kara River; d – 1x2 m relict of a coal fragment (black, in
 830 the center) within suevite, from the right bank of the Kara River. Insets: 2a, c – thin polished
 831 sections (2 cm diameter) in transmitted light, numbers correspond to relict fragments: 1 – coal , 2

832 –sandstone, 3 – limestone, 4 – black shale; 5 – pore; 2b – coal pieces from a large coal fragment.

833 The color image is available within an on-line version.

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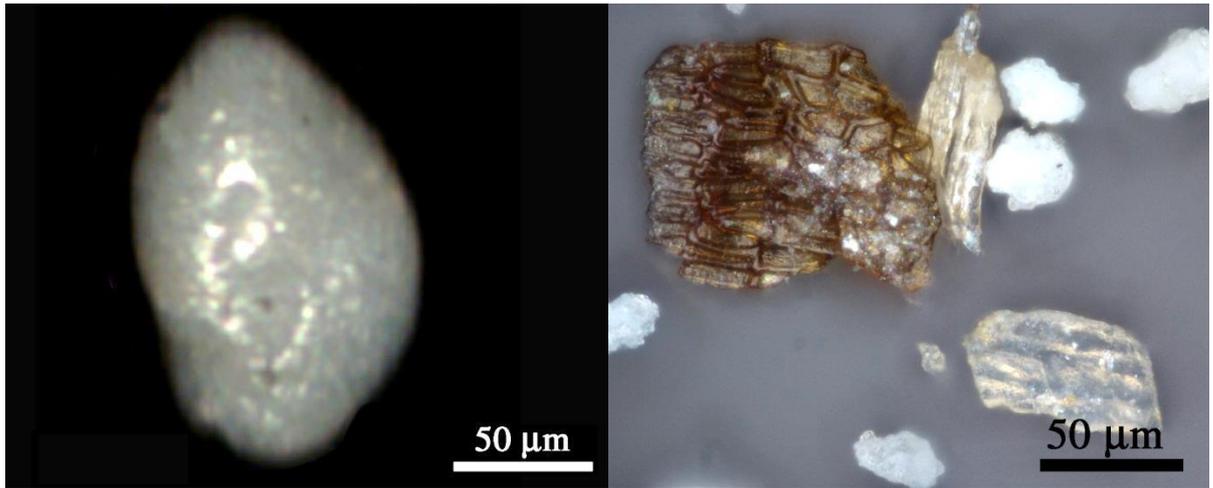
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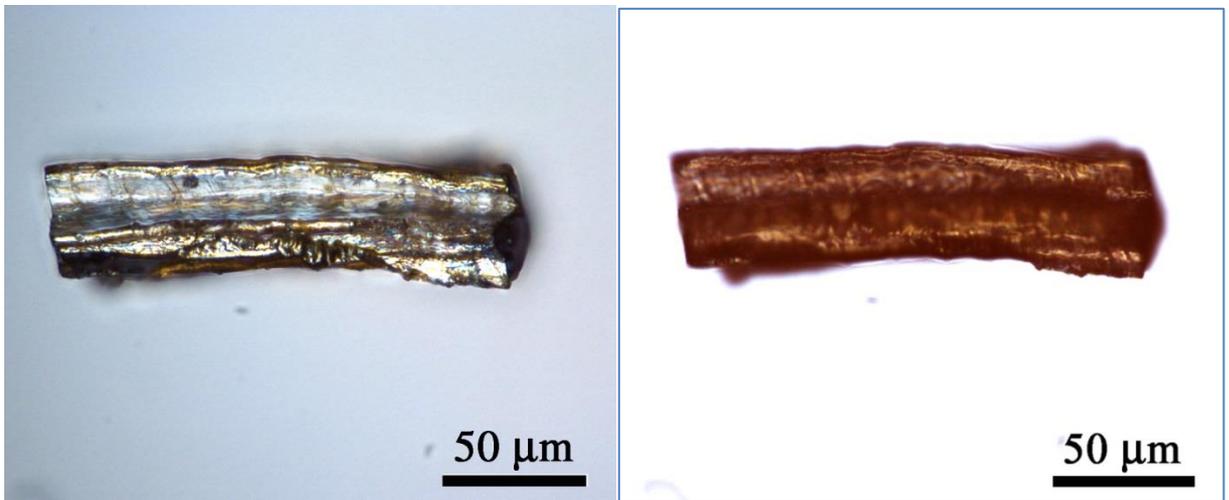


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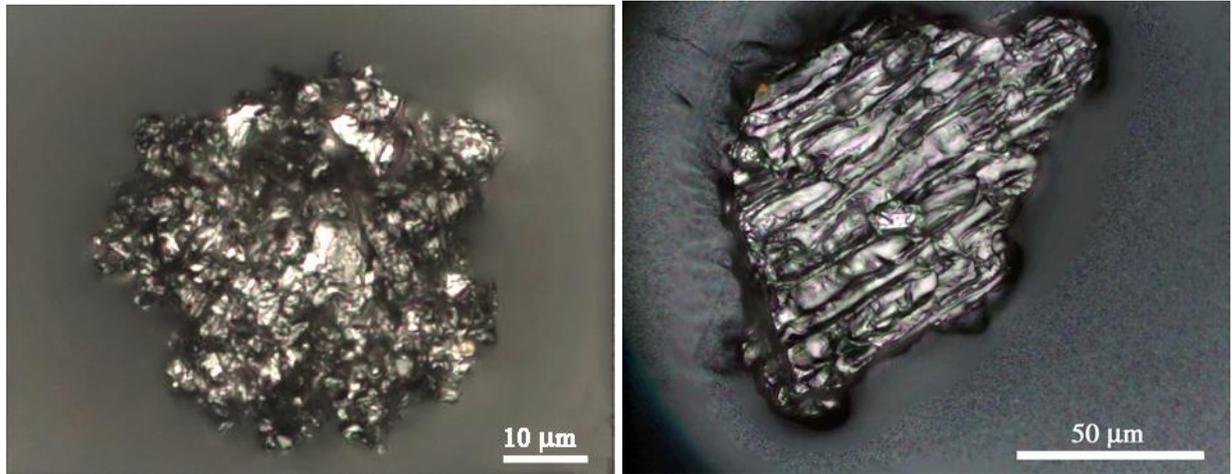


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Fig. 3. Optical images of the varieties of impact after-coal carbons obtained from

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thermochemical concentrate in unpolarized reflected light: a – Light gray, fully crystalline, dense

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sugar-like diamond with a rounded egg-like shape, in reflected light. A SEM image of the same

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grain is shown in fig. 4a; b – After-organic diamond paramorphs and friable sugar-like particles.

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A diamond paramorph after-organic relict (d – shown in transparent light); e – hedgehog-shaped

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graphite; f – GLC paramorph after organics. The color image is available within an on-line

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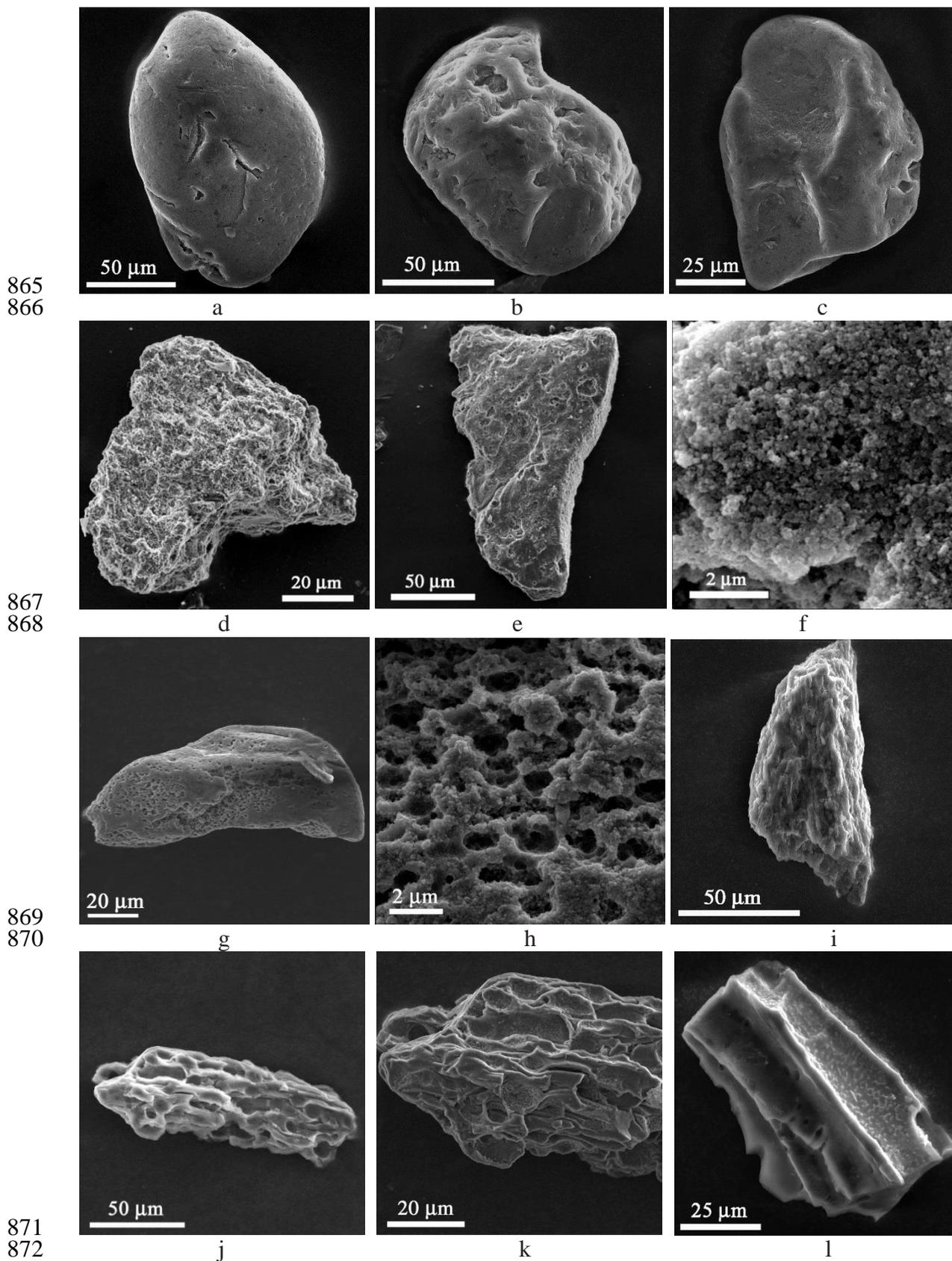
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873 Fig. 4. SEM images of varieties of impact after-coal diamonds. a-f – Sugar-like diamonds.
874 Dense rounded egg-like grain with deep cracks (a); dense rounded drop-like grains (b); dense
875 irregular grain with rounded edges and smooth surface (c); white friable sugar-like particle (d);

876 white friable sugar-like particle (e) with a magnified fragment (f). Irregular sugar-like dense
877 diamond with microporous structure (g) and a close-up (h). Brown poorly ordered after-organic
878 diamond paramorphs: i – diamond with fibrous morphology; j – After-organic paramorph with
879 negative relict morphology after framboidal pyrite aggregates and close-up (k). l – Diamond
880 paramorph with tiny relict organic morphology.

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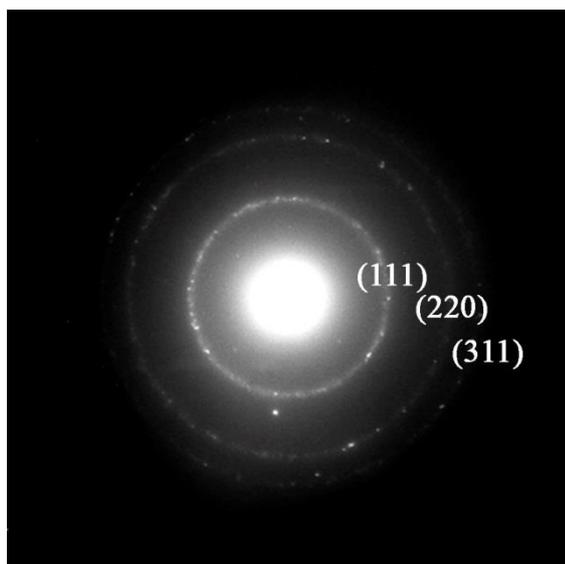
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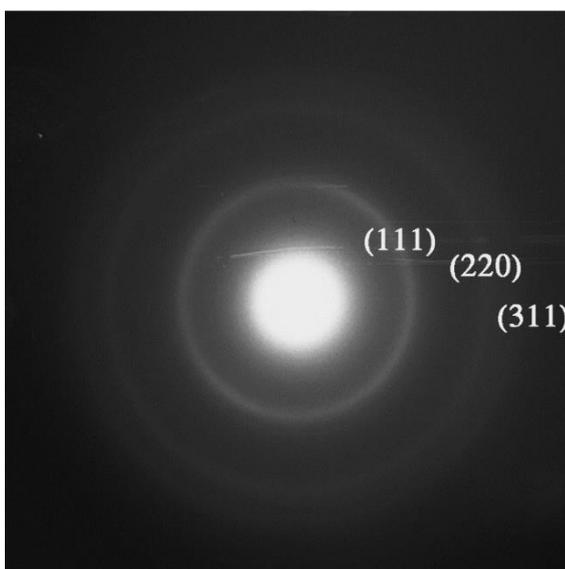
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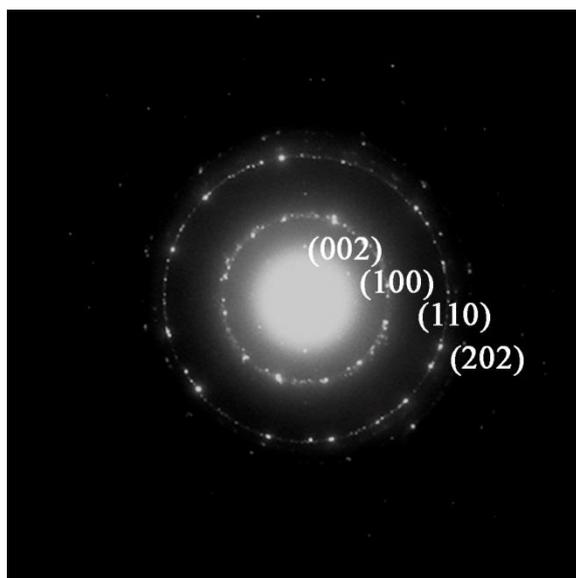
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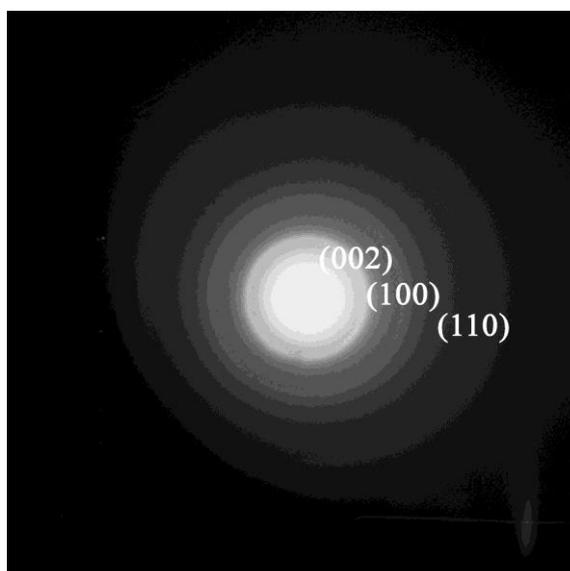
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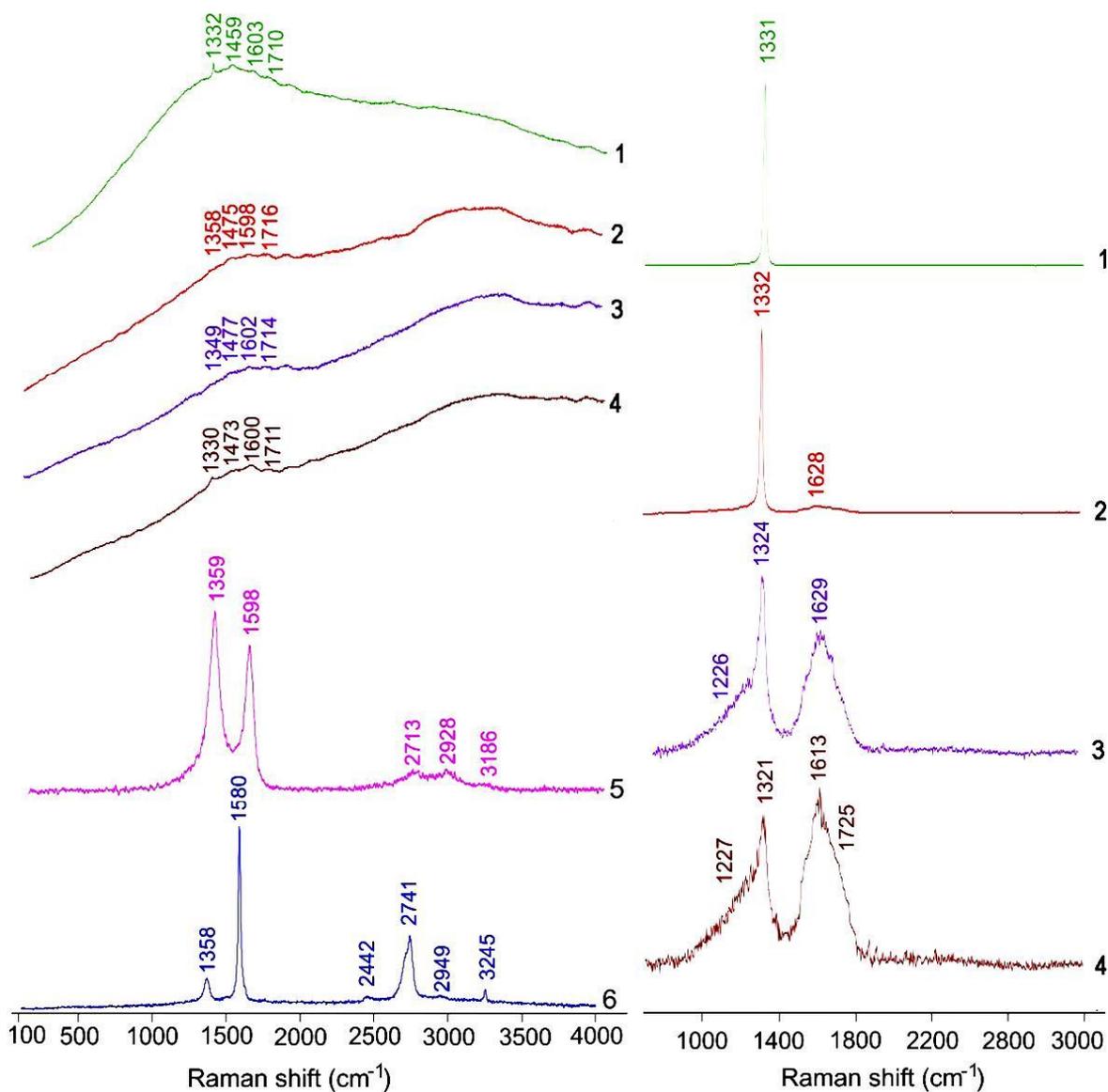
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908 Fig. 5. Typical transmission electron microscopy data of Kara impact after-coal carbons,
 909 electron diffraction patterns: a – sugar-like diamond $d_{111}= 2.06 \text{ \AA}$, $d_{220}= 1.27 \text{ \AA}$, $d_{311}= 1.07 \text{ \AA}$; b –
 910 nanopolycrystalline diamond, $d_{111}= 2.05 \text{ \AA}$, $d_{220}= 1.25 \text{ \AA}$, $d_{311}= 1.06 \text{ \AA}$; c – polycrystalline
 911 graphite, $d_{100}=2.11 \text{ \AA}$, $d_{110}=1.23 \text{ \AA}$; $d_{112}=1.155 \text{ \AA}$; d – GLC, diffusion rings with maxima at
 912 $d_{002}=3.38 \text{ \AA}$, $d_{100}=2.07 \text{ \AA}$, $d_{110}=1,23 \text{ \AA}$. The phases identification has been supported by Raman
 913 spectroscopy measurements.

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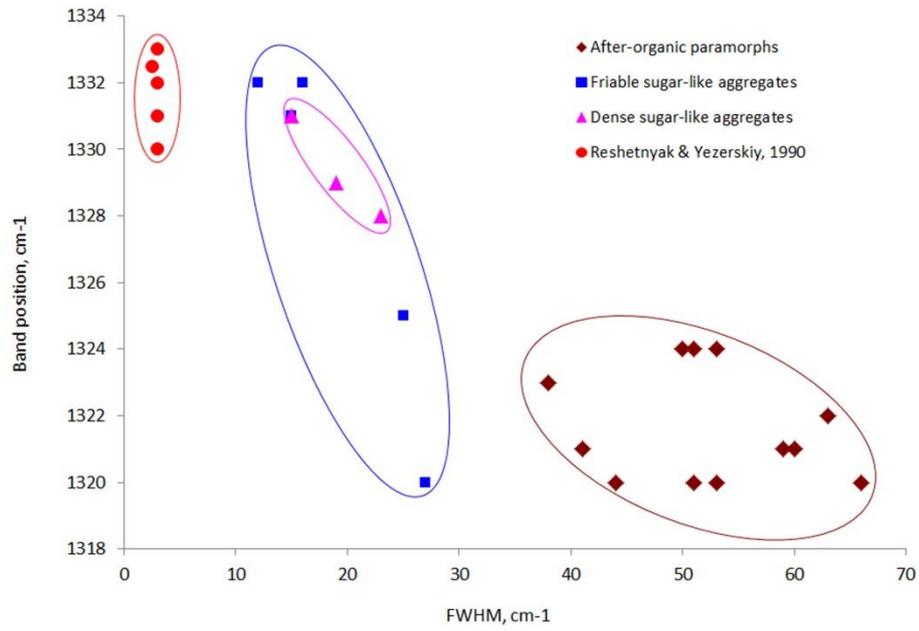
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918 Fig. 6. Raman spectra for different impact after-coal carbons. Spectra on the left are for visible
 919 laser excitation, and on the right for UV laser excitation. Spectra of micrograined sugar-like
 920 well-ordered white diamonds: 1 – dense, 2 – friable. Spectra 3 and 4 are from less-ordered
 921 brown diamonds, less ordered with different crystallinity, with 3 showing better crystallinity than
 922 4. Spectrum 5 is from glass-like carbon. Spectrum 6 is from graphite. The color image is
 923 available within an on-line version.



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925 Fig. 7. Raman spectroscopy data of after-coal diamond varieties for A_{2g} diamond mode. The
 926 color image is available within an on-line version.

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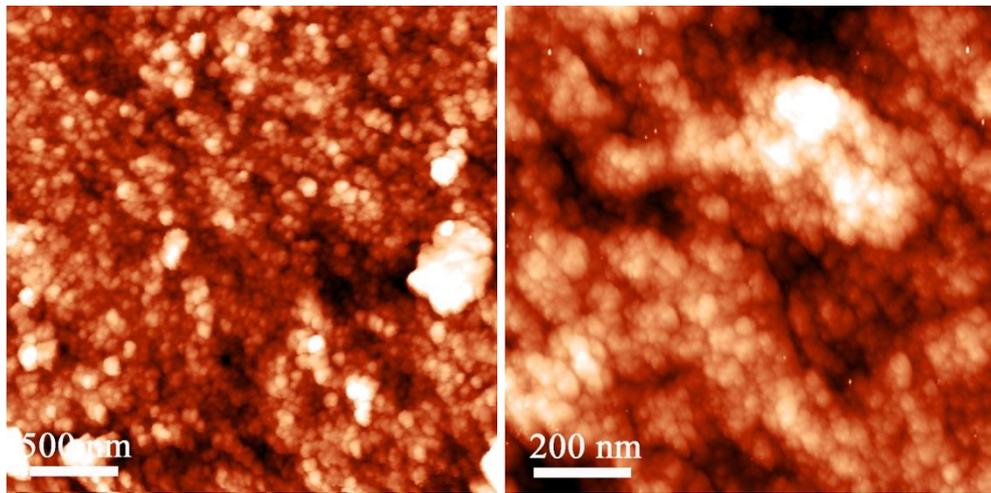
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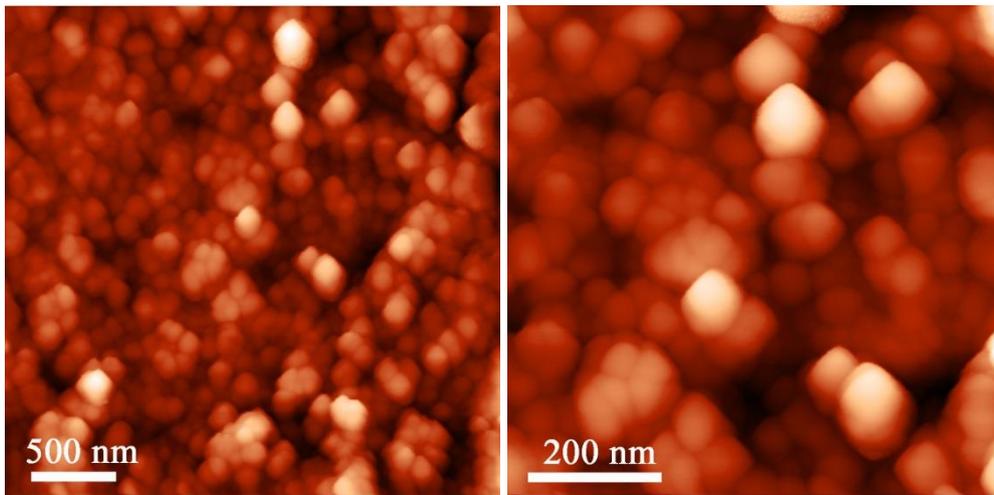
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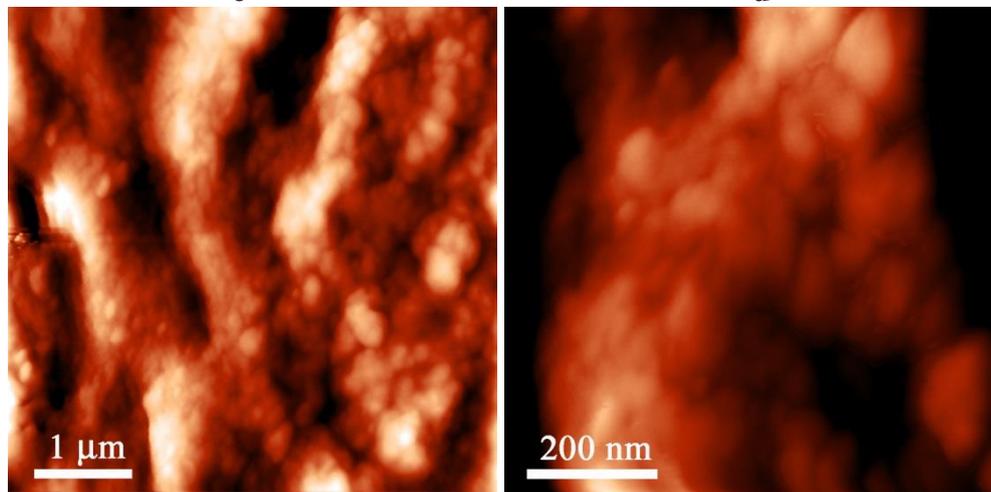
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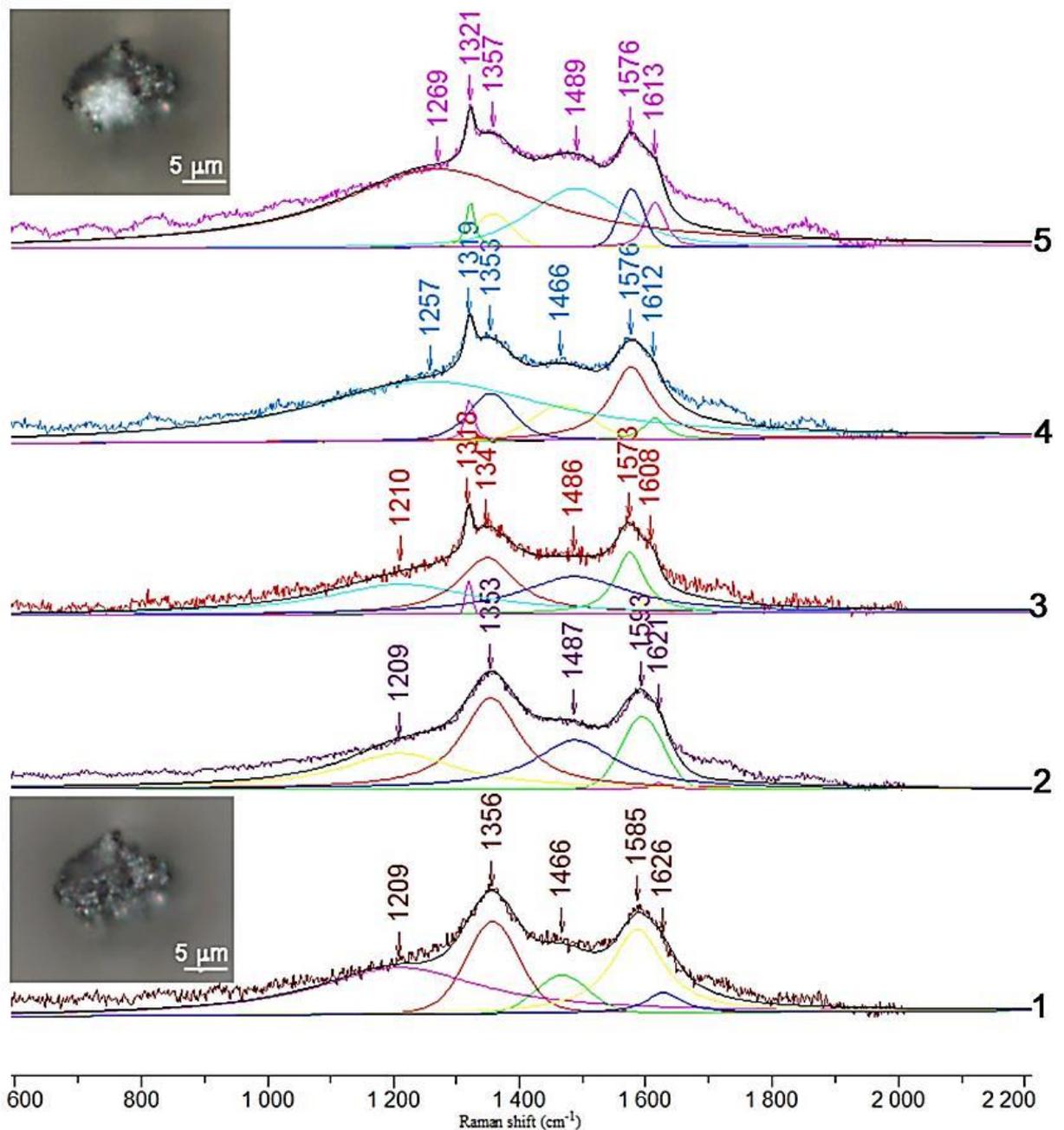
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944 Fig. 8. Nano-scale features of after-coal impact diamonds by AFM data: a, b – White sugar-like
945 dense diamond of irregular shape consisting of hierarchical diamond nanocrystal aggregates; c, d
946 – Brown after-organic paramorph with fine-scale but well-defined octahedral nanocrystals; e, f –
947 After-organic diamond paramorph with poorly recognizable nanocrystal morphology. The color
948 image is available within an on-line version.



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950 Fig. 9. Raman study of coated after-coal diamond using visible laser excitation (488 nm). 1-5
 951 continually measured Raman spectra (1 - starting, 5 – final). Spectra 1 and 2 are at 1.2 mW laser
 952 power, shown along with corresponding photo inset at the bottom. Spectra 3-5 are at 12 mW.
 953 The corresponding upper photo inset reveals the recovered diamond as a bright central spot after
 954 the 12 mW laser heating treatment. The colored curves are fitting of the spectra to components
 955 with upper set band positions. The color image is available within an on-line version.

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